

THE SYNTHESIS OF 4',5,6,7- AND 4',5,7,8-TETRAHYDROXYFLAVANONE
AND THEIR COMPARISON WITH CARTHAMIDIN AND ISOCARTHAMIDIN

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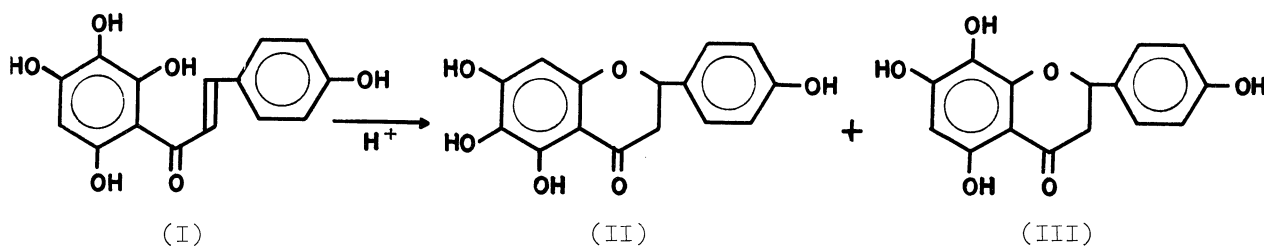
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4',5,6,7- and 4',5,7,8-Tetrahydroxyflavanone (II and III)
have been synthesized by the isomerization of 2',3',4,4',6'-
pentahydroxychalcone (I), and the structures of carthamidin
and isocarthamidin were corrected by the comparison with these
synthetic flavanones.

In a previous communication,¹⁾ we have reported the synthesis of 2',3',4,4',6'-
pentahydroxychalcone (I) corresponding to the aglycon of carthamin,²⁾ the red
coloring matter of the flowers of Carthamus tinctorius. In the present paper, the
synthesis of 4',5,6,7- and 4',5,7,8-tetrahydroxyflavanone (II and III) by the
isomerization of I with acid, and their comparison with carthamidin and isocartha-
midin, the hydrolysis products of carthamin²⁾ will be described.

A mixture of II and III was obtained from I by heating with dilute hydrochloric
acid in methanol, and the mixture was chromatographed on a column of silica gel to
give II and III in 32 and 29% yields respectively.

The structures of II and III were confirmed by the elemental analysis, by the
studies of their UV, IR, and NMR spectra, and, furthermore, by the comparison of
their tetramethylethers with authentic samples of 4',5,6,7-³⁾ and 4',5,7,8-tetra-
methoxyflavanone⁴⁾, respectively.



4',5,6,7-Tetrahydroxyflavanone (II), mp 219°C (pale yellow needles from dilute methanol); Mg-HCl, deep red; FeCl₃, brownish green; UV, $\lambda_{\text{max}}^{\text{EtOH}}$ (log ϵ) 300 (4.13) and 368 m μ (3.18); tetraacetate, mp 160°C (colorless prisms from methanol).

4',5,7,8-Tetrahydroxyflavanone (III), mp 248°C (pale yellow needles from methanol); Mg-HCl, deep red; FeCl₃, brownish green; UV, $\lambda_{\text{max}}^{\text{EtOH}}$ (log ϵ) 299 (4.17) and 370 m μ (3.52); tetraacetate, mp 177°C (colorless needles from methanol).

While Kuroda²⁾ had been reported the structure of III and II for carthamidin (A) and isocarthamidin (B) on the basis of the color phenomena of their hydroxyl groups, the IR spectra of A and B which were prepared by the hydrolysis of carthamin according to Kuroda's method²⁾ were identical with those of the synthetic II and III.

Seshadri et al³⁾ have reported the synthesis of II and III by the demethylation of the corresponding tetramethoxyflavanones in the presence of anhydrous aluminium chloride in benzene solution. We have attempted the synthesis in a manner similar to that described above, but, no expected tetrahydroxyflavanones could be obtained.

The detail will be published elsewhere.

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